

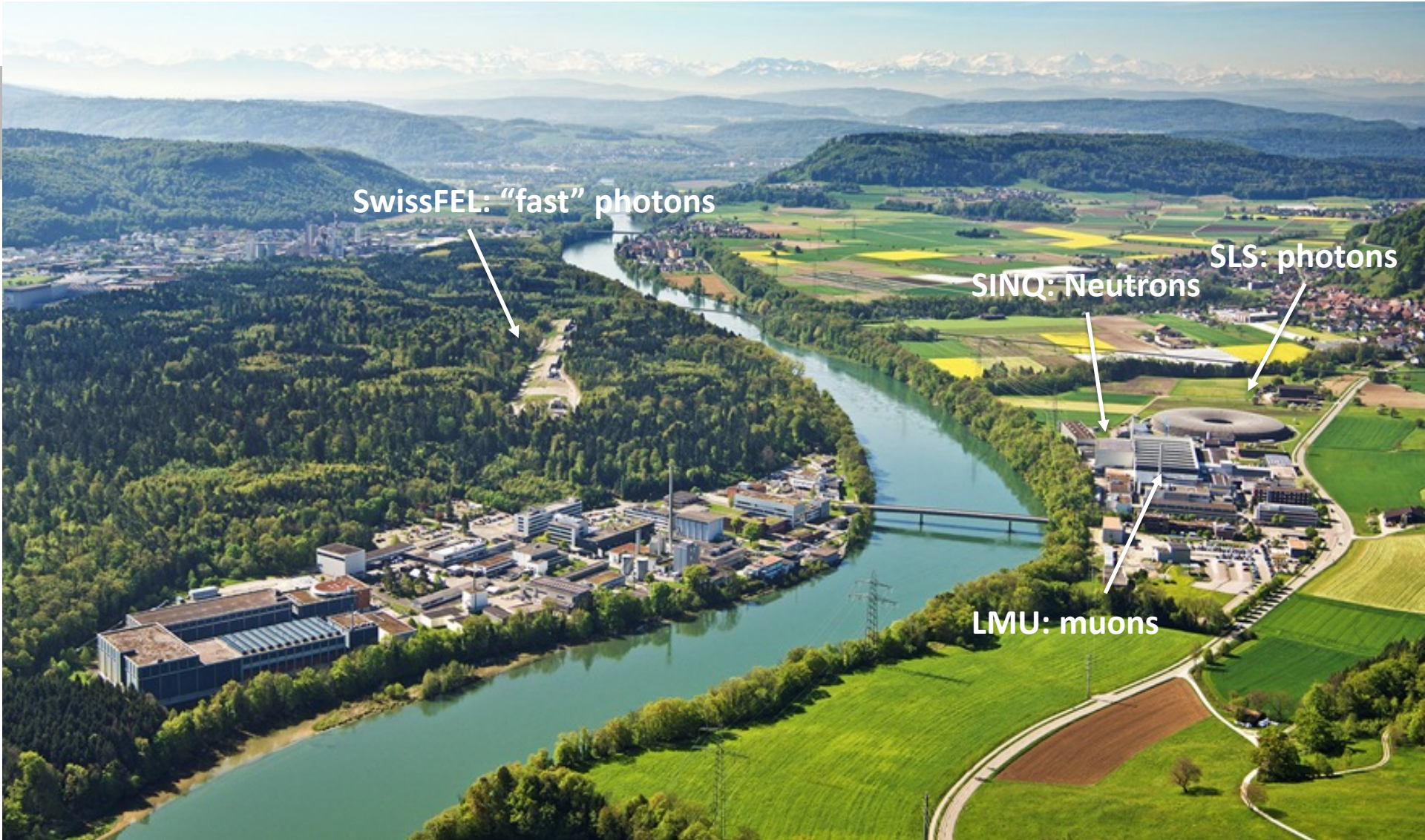


Wir schaffen Wissen – heute für morgen

**Angle-resolved photoemission
spectroscopy,
the microscope for the electronic
structure**

*Milan
Radovic*

Large scale facilities at PSI: the unique toolset!



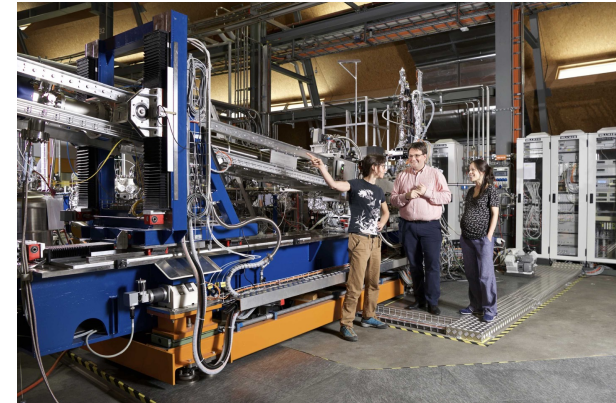
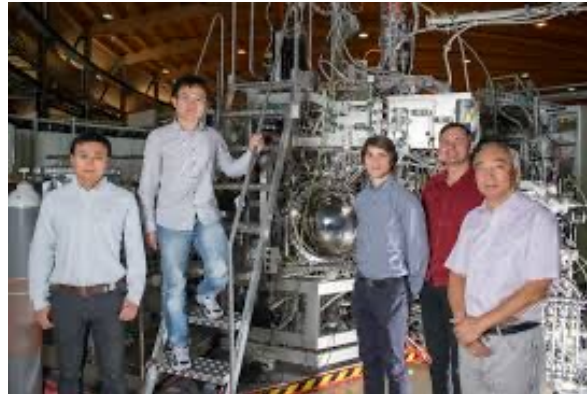
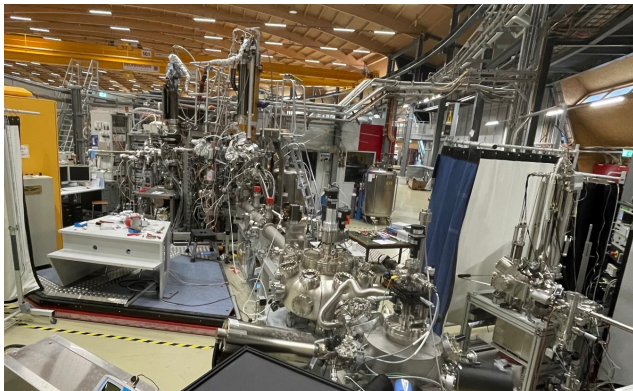
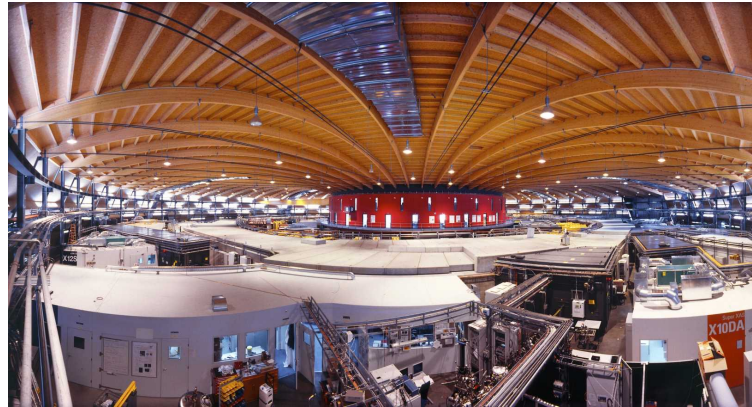
SwissFEL: "fast" photons

SINQ: Neutrons

SLS: photons

LMU: muons

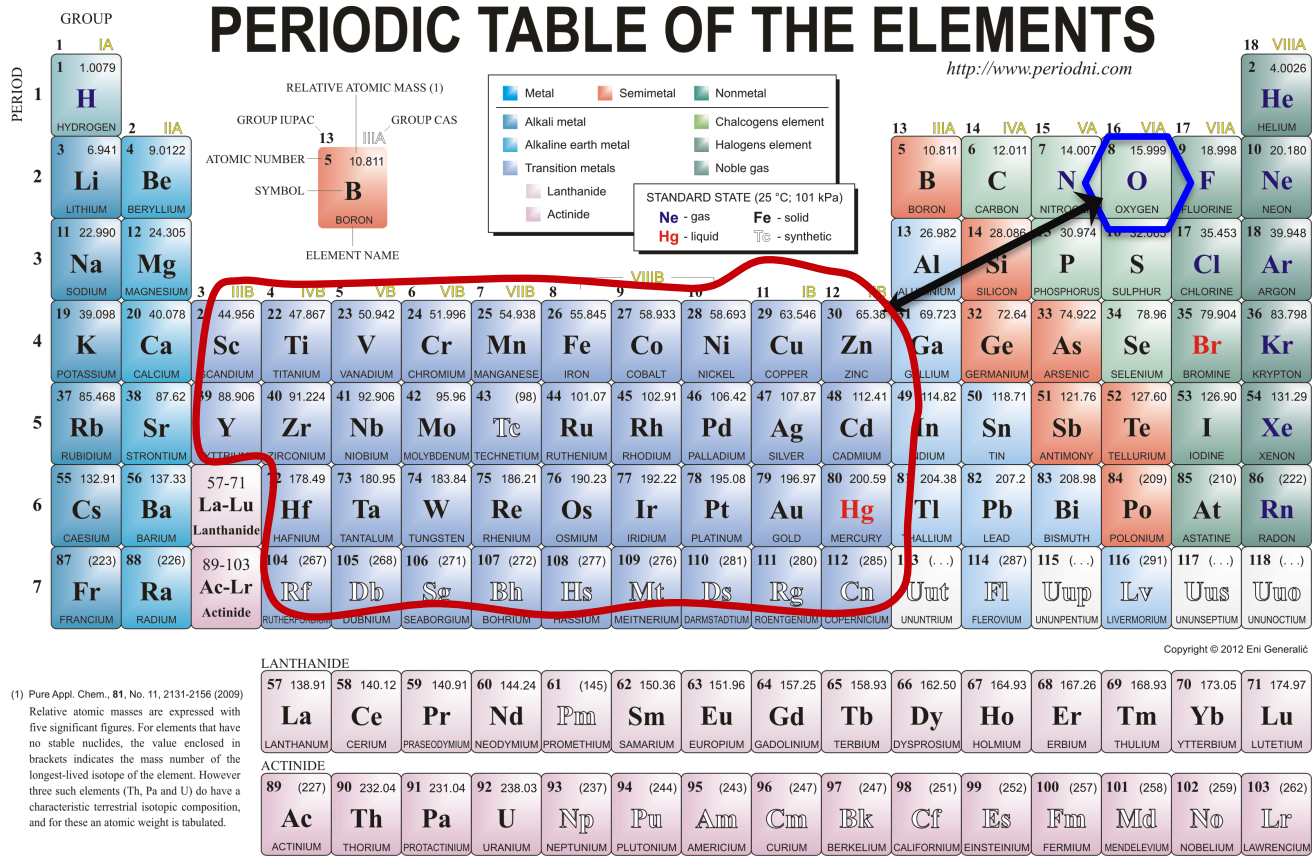
Swiss Light Source



Outline

- The general introduction of Transition Metal Oxides (TMOs)
- Band structure: basic concept
- Angle-resolved photoemission spectroscopy –ARPES:
overview
- Example: Depicting the electronic structure of Low
Dimensional Electronic Structure at ATiO_3

Creating novel PHASES and Tuning of ELECTRONIC states of artificial and hybrid materials based on TMOs

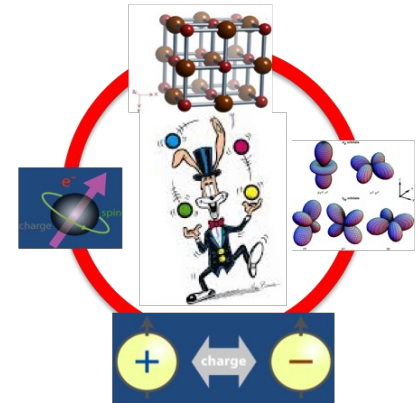


(1) Pure Appl. Chem., 81, No. 11, 2131-2156 (2009)
Relative atomic masses are expressed with five significant figures. For elements that have no stable nuclides, the value enclosed in brackets indicates the mass number of the longest-lived isotope of the element. However three such elements (Th, Pa and U) do have a characteristic terrestrial isotopic composition, and for these an atomic weight is tabulated.

TMO show extraordinary electronic and magnetic properties: CMR, HTc, MIT, etc.
Transition Metal Oxides: Surface Chemistry and Catalysis H.H. Kung, Elsevier Science, NY, 1989.

What are Transition metal Oxides (TMO)?

- **Partly filled d-shell: electron-electron interactions, thus spin and orbital degrees of freedom play role.**
- **Multiple valence states: many electronic configurations.**
- **Easy charge exchange with Oxygen.**



Many structures possible, very different properties.

- **BO Rocksalt structure: NiO, ZnO, TiO, CoO...**
- **ABO₃ Perovskite: cubic SrTiO₃, Orthorombic: ReNiO₃, SrIrO₃, Tetragonal: TiO₂, BaTiO₃**
- **A₂BO₄ Perovskite (e.g La₂CuO₄): Layered structure (weak interplane coupling),**
- **Ruddlesden-Popper series A_{n+1}B_nO_{3n+1} (Interpolates between 2 and 3 dimensional coordination: LaSr₃FeO₁₀, La₃Ni₂O₇ ...**
- **Double perovskite AA'BB'O₆ (Sr₂FeMoO₆)**
- ...

Creation and Control of the electronic properties of ABO₃ Perovskite

Distortion of the BO₆ octahedron



Energy splitting of the
 d_{xy} & d_{xz} & d_{yz} bands

Doping (through A or O vacancies)



Filling of the bands

Octahedral rotations, Binding angles



Hopping probability,
Effective mass
Band width

Proximity effects

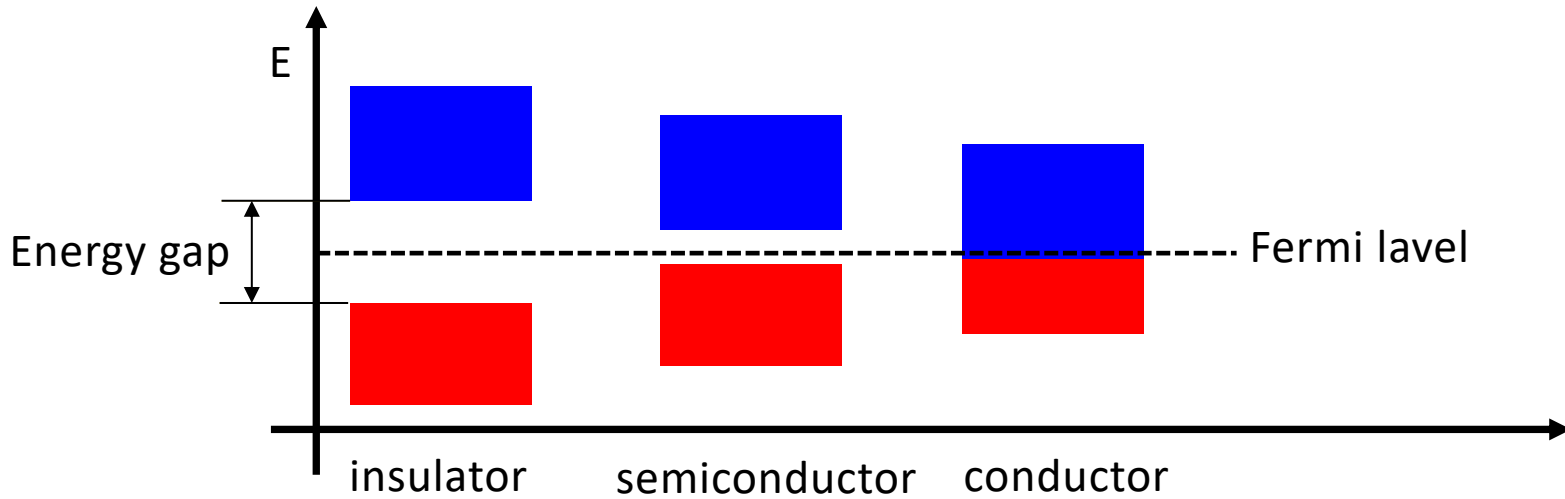
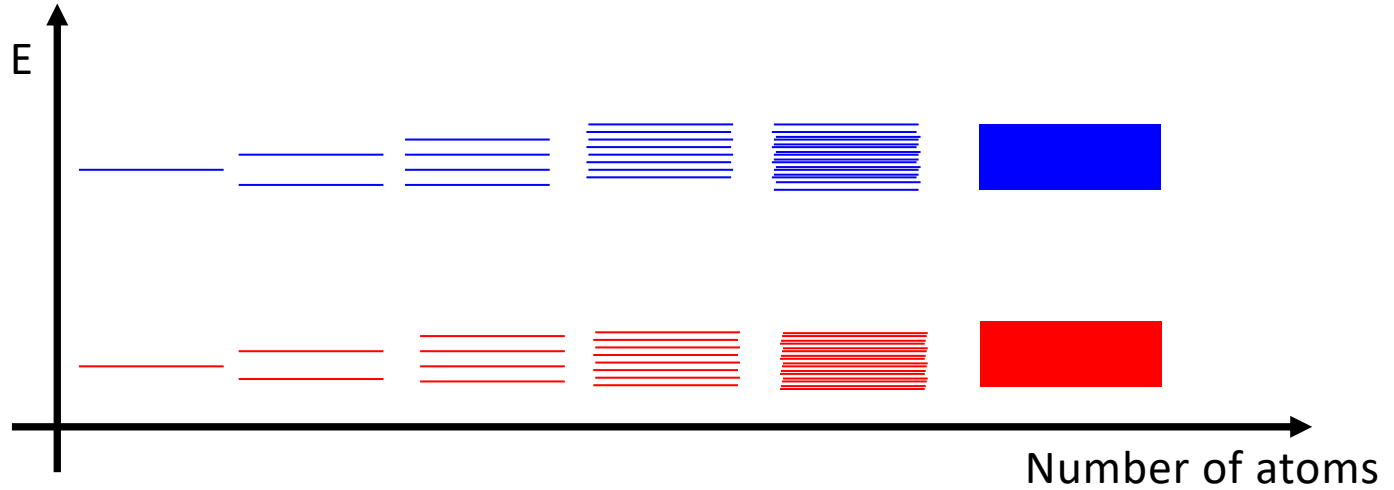


Doping via charge transfer,
Inducing orders
(magnetism
superconductivity...)

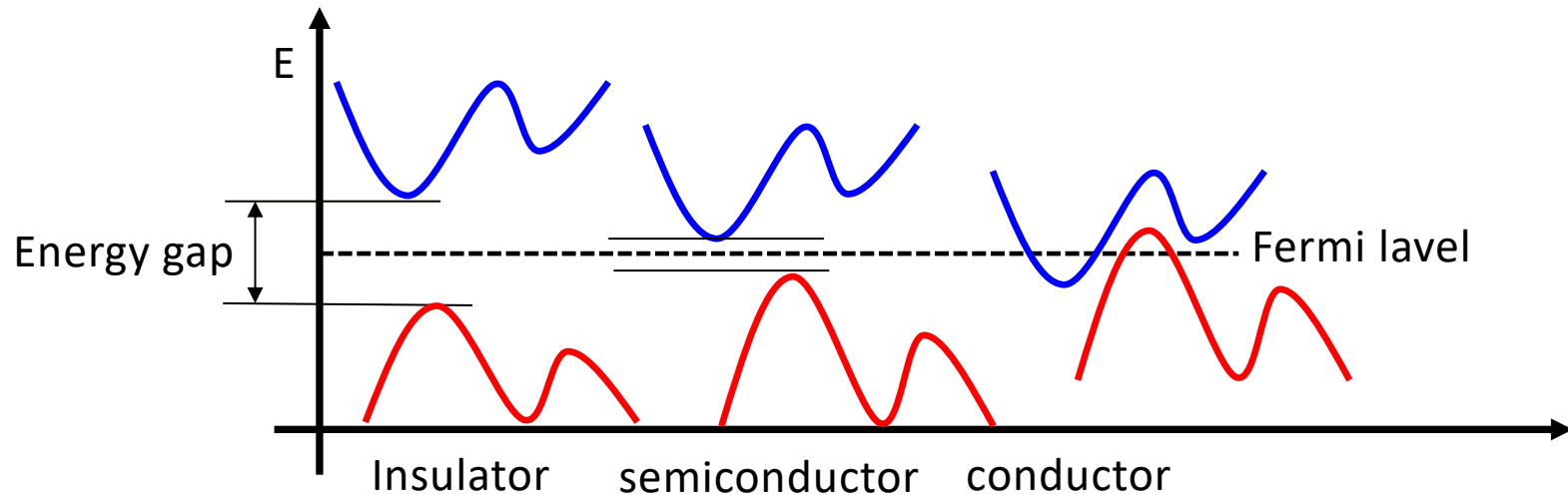
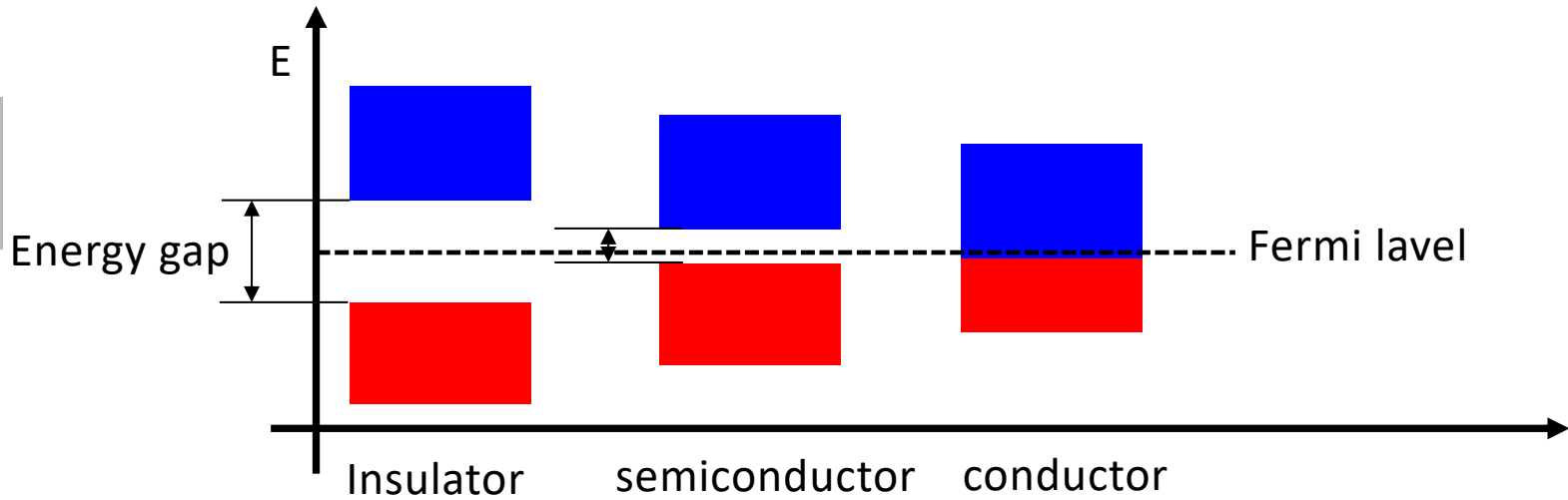
How to turn the knobs?

- ✓ Temperature (phase transition);
 - ✓ Defects;
 - ✓ Cristal structure;
 - ✓ *lattice strain, proximty...*
- ✓ Stacking (artificial VdW materials)

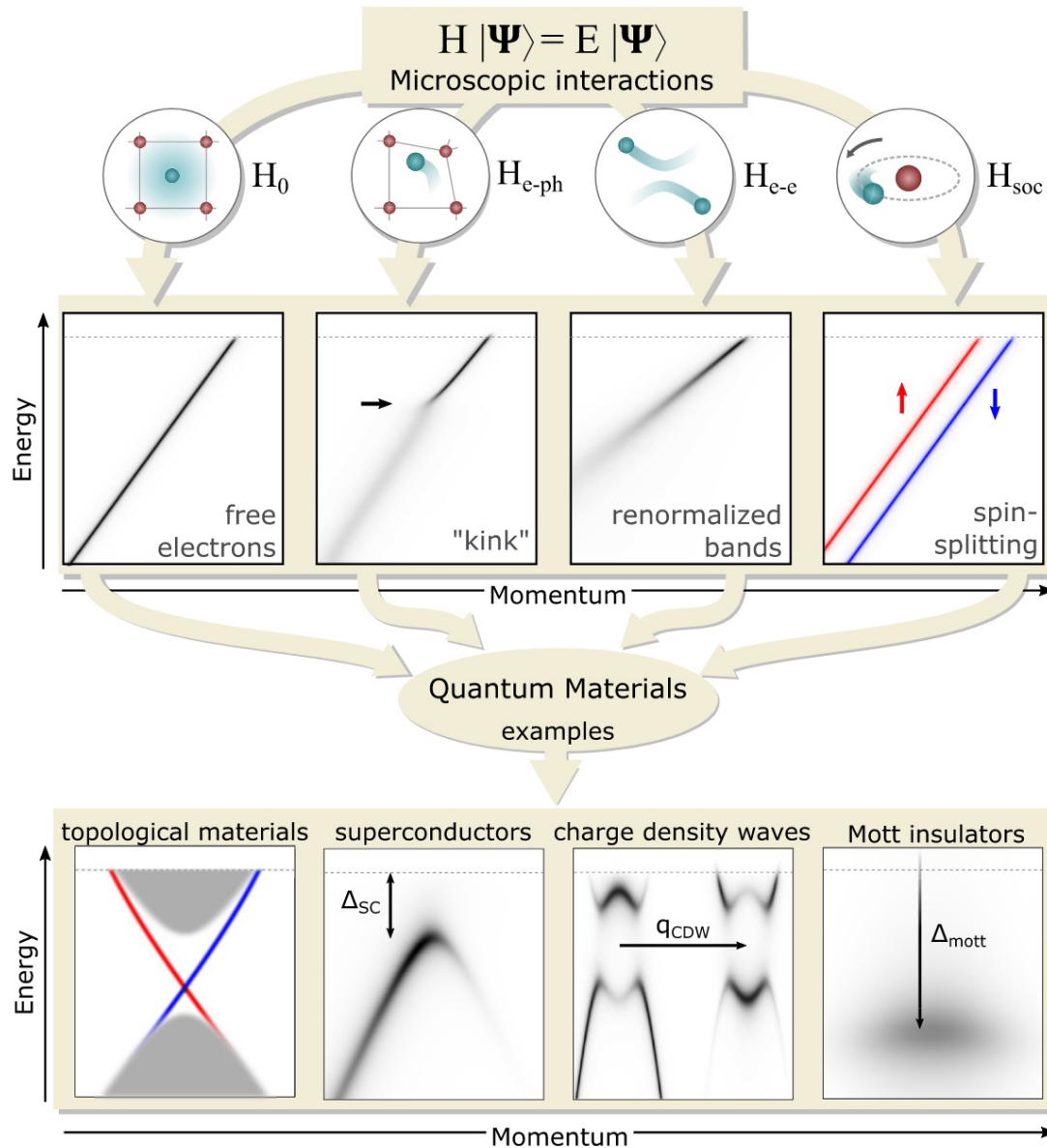
Band structure



Band structure



Band structure of quantum materials



Angle-resolved photoelectron spectroscopy: the microscope for the electronic structure

Two important parameters of electrons in a solid

Bound to the lattice

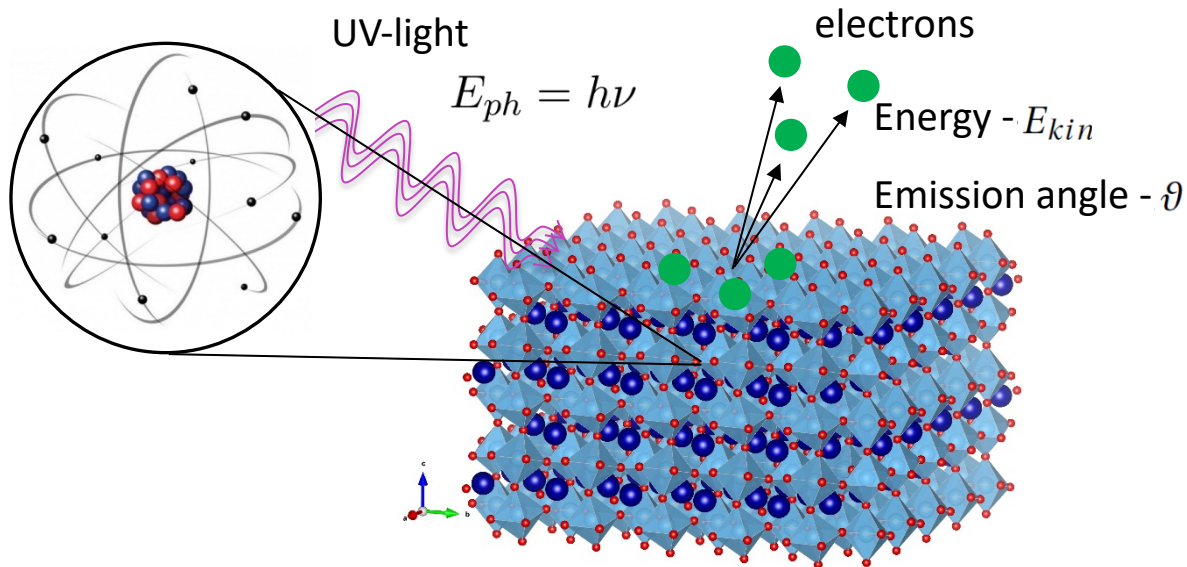
→ binding energy

$$E_{bin}$$

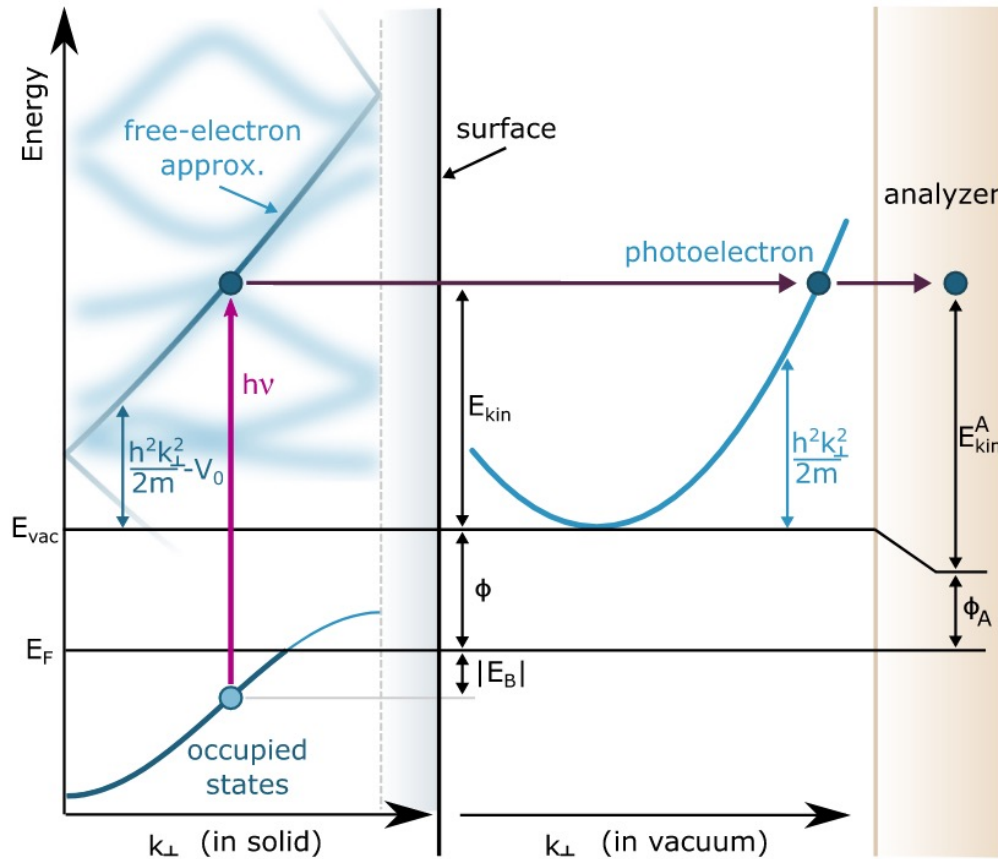
Movement with velocity \vec{v}

→ momentum

$$\vec{k} = m\vec{v}/\hbar$$



Angle-resolved photoelectron spectroscopy: the microscope for the electronic structure



Electron energy \rightarrow binding energy in the crystal:

$$E_{bin} = h\nu - E_{kin} - \Phi_A$$

Emission angle \rightarrow in-plane momentum:

$$k_{\parallel} = \sqrt{\frac{2m_e}{\hbar^2} E_{kin}} \sin \vartheta$$

Changing photon energy \rightarrow out-of-plane momentum:

$$k_{\perp} = \frac{1}{\hbar} \sqrt{2m_e(E_{kin} \cos^2 \vartheta + V_0 + \Phi_A)}$$

Emission angle - ϑ

A bit more

ARPES intensity:

$$I(\mathbf{k}, \omega) = I_0(\mathbf{k}, h\nu, \mathbf{A}) A(\mathbf{k}, \omega) f(\omega)$$

$$I_0 \sim M_{f,i}^k = \langle \phi_{\bar{f}}^k | \mathbf{A} \cdot \mathbf{p} | \phi_i^k \rangle.$$

proportional to the square of the dipole matrix element

$$A(\mathbf{k}, \omega) = -\frac{1}{\pi} \frac{\Sigma''(\mathbf{k}, \omega)}{[\omega - \varepsilon_0(\mathbf{k}) - \Sigma'(\mathbf{k}, \omega)]^2 + [\Sigma''(\mathbf{k}, \omega)]^2}$$

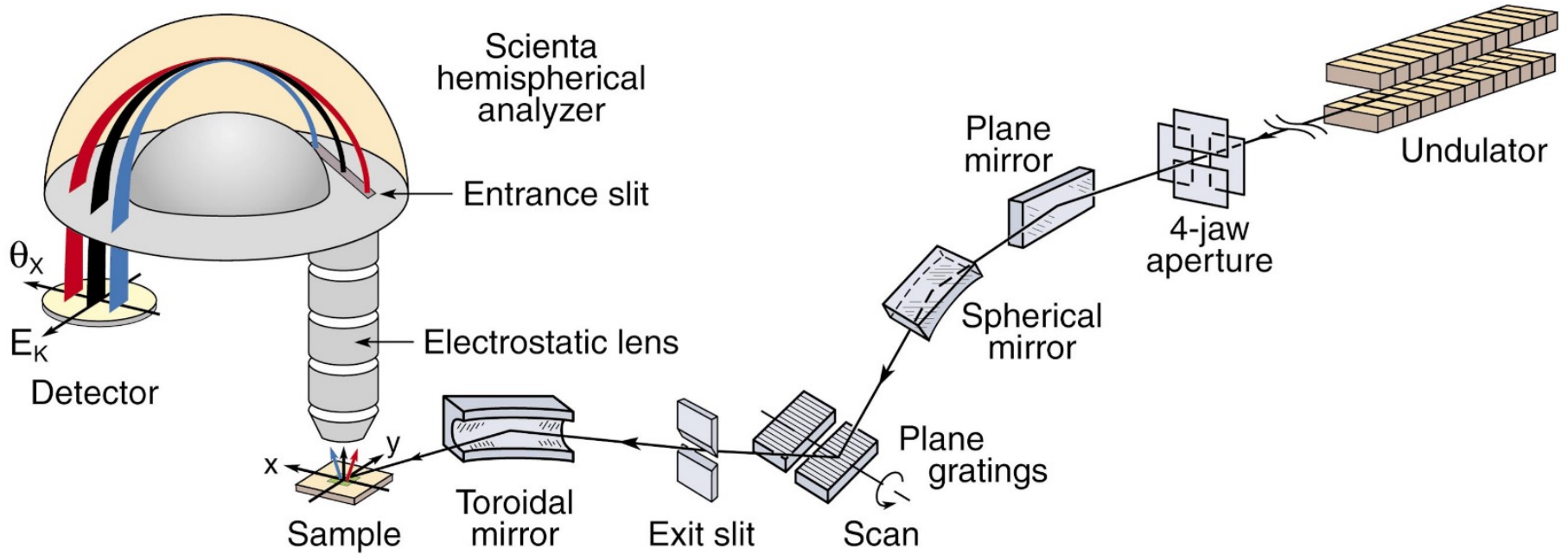
the single-particle spectral function:
electronic band structures and lifetimes

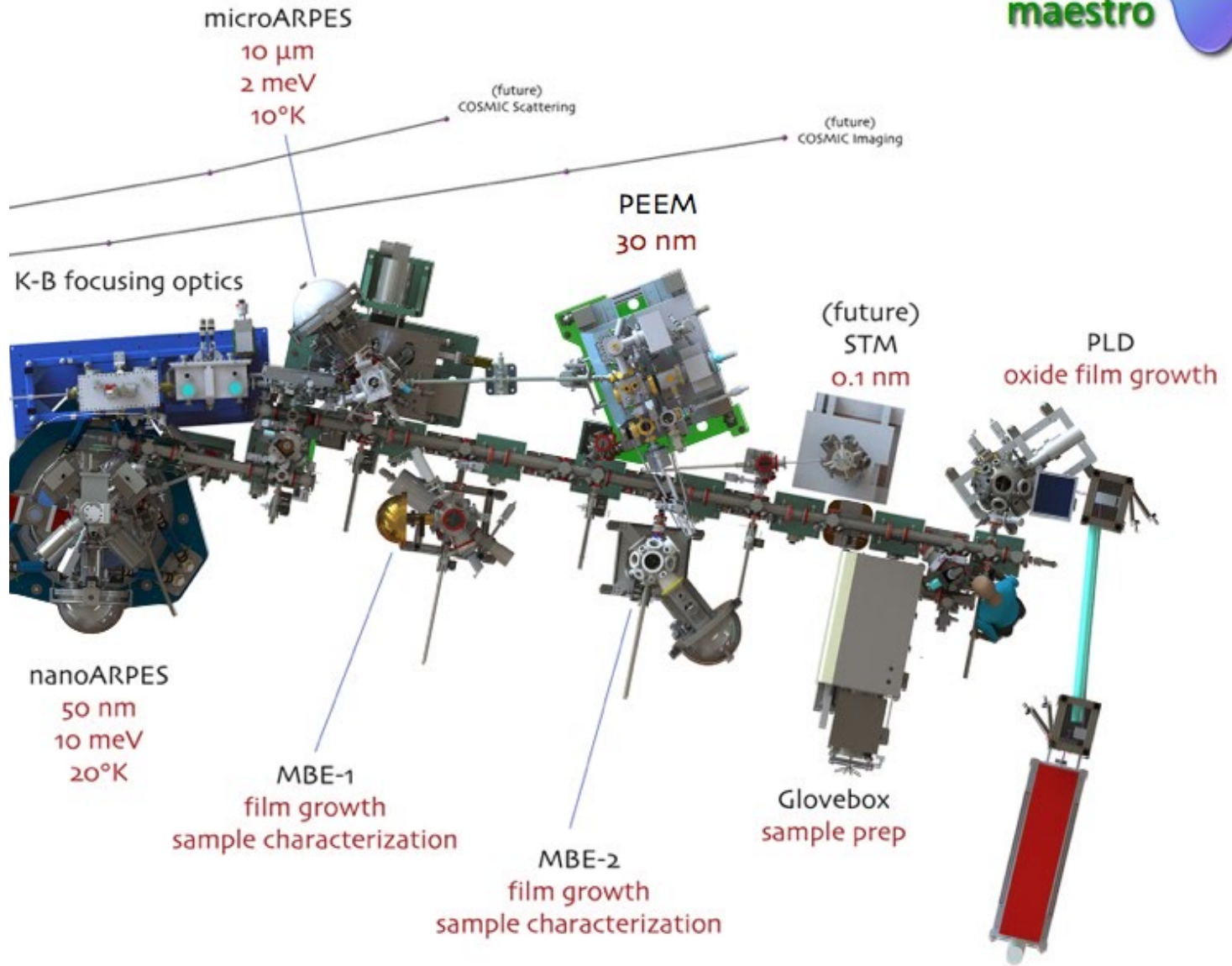
$\varepsilon_0(\mathbf{k})$: the non-interacting (bare)
energy-band dispersion

$$f(\mathbf{k}, \omega) = 1 / (e^{\omega/k_B T} + 1),$$

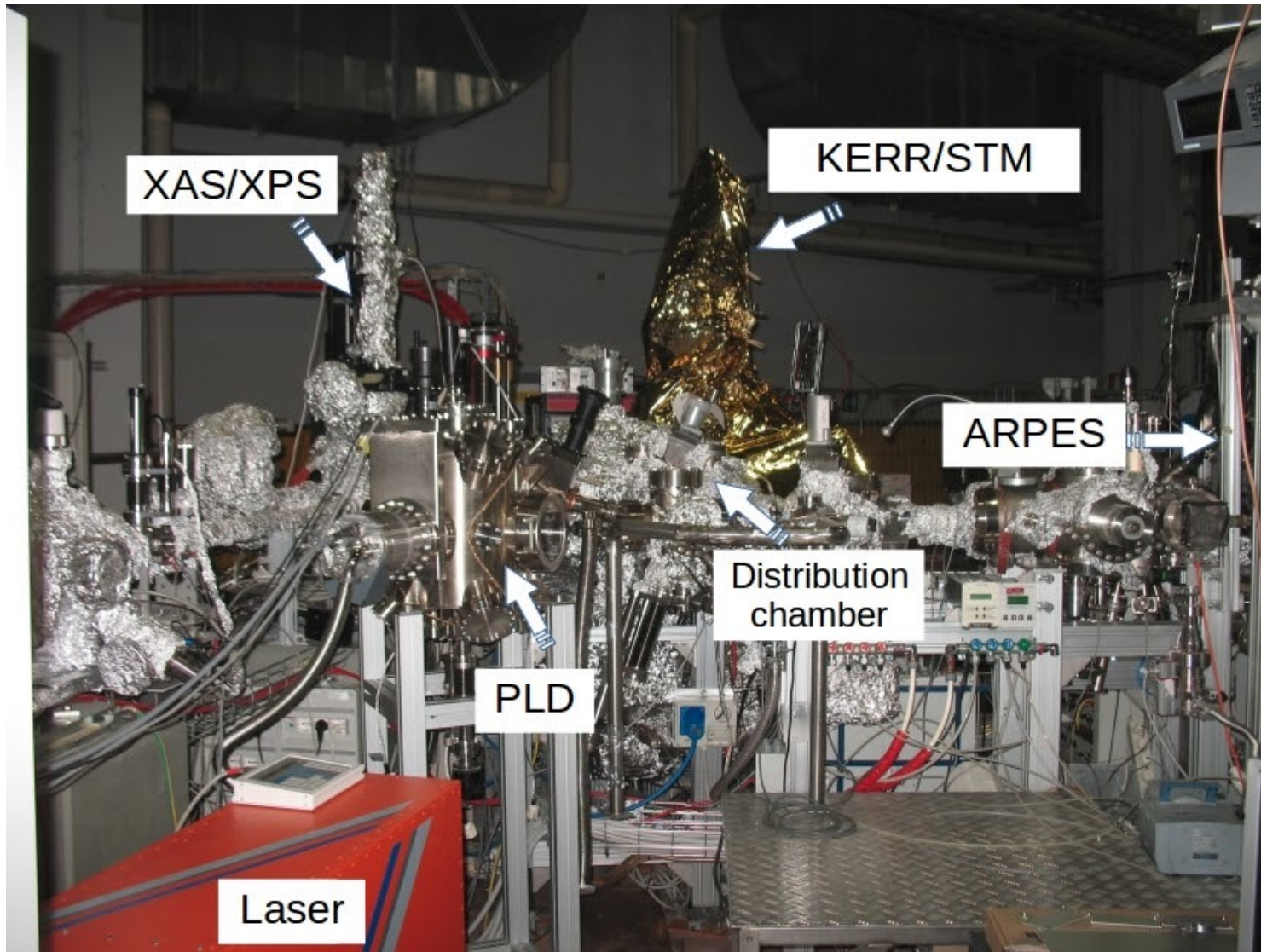
Fermi–Dirac function

Angle-resolved photoelectron spectroscopy: the microscope for the electronic structure.



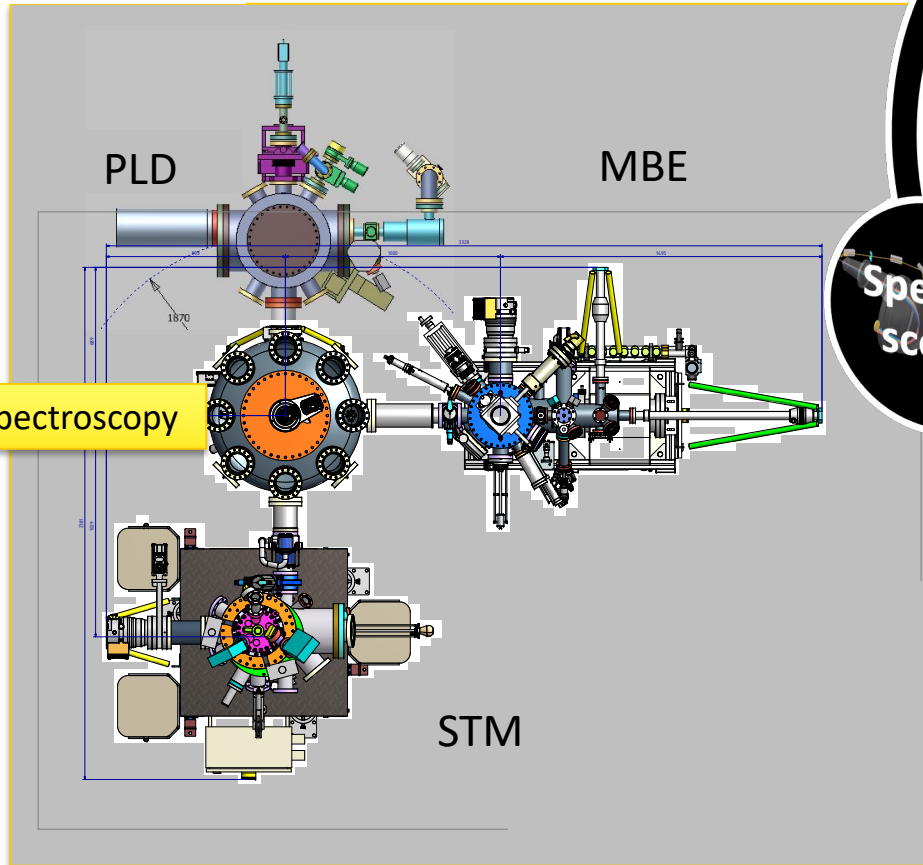
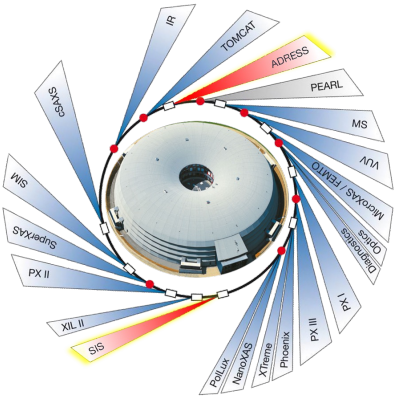


PLD + ARPES at APE beamline, Elettra (Italy)

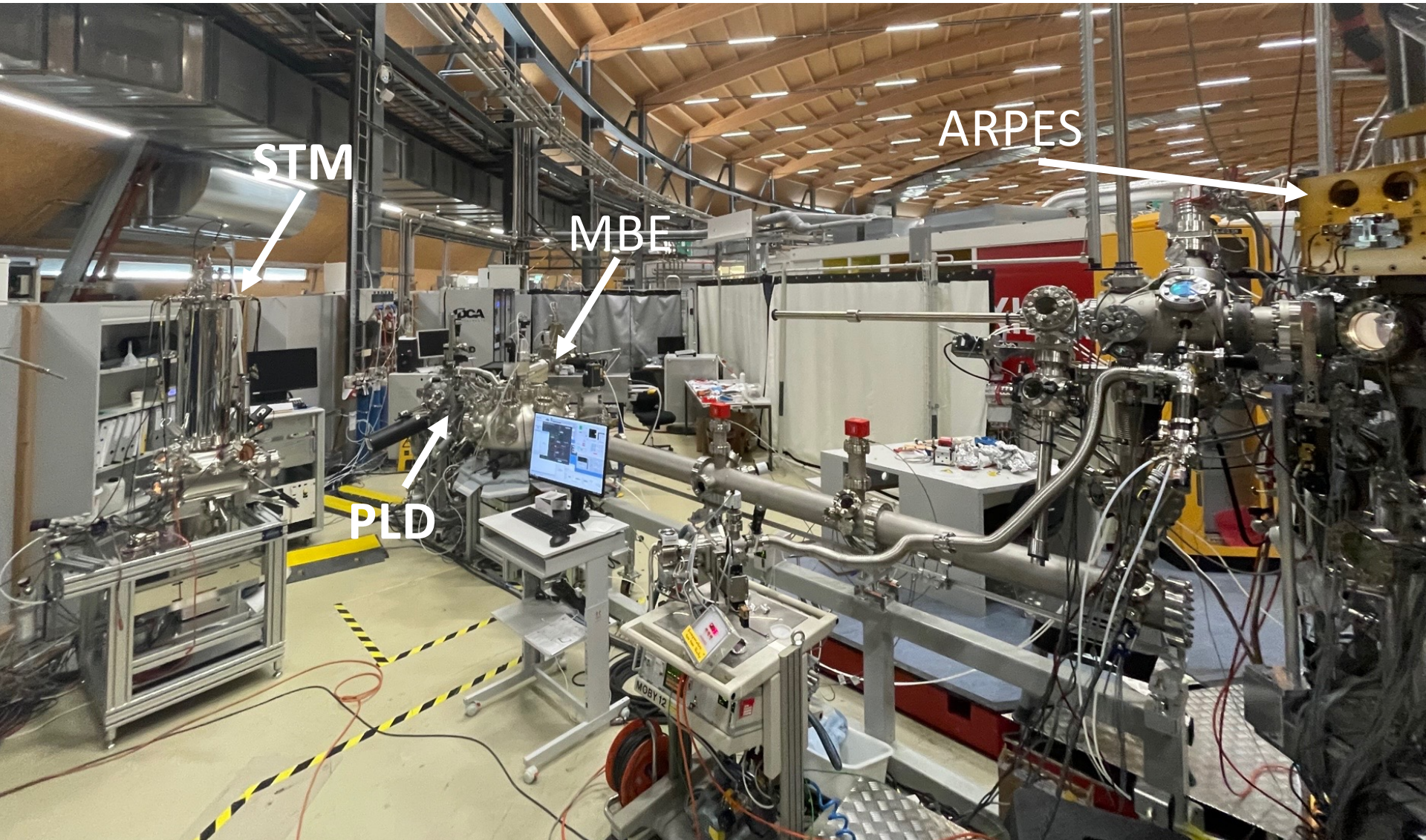


The modular system at the SIS beamline: PLD+STM+MBE

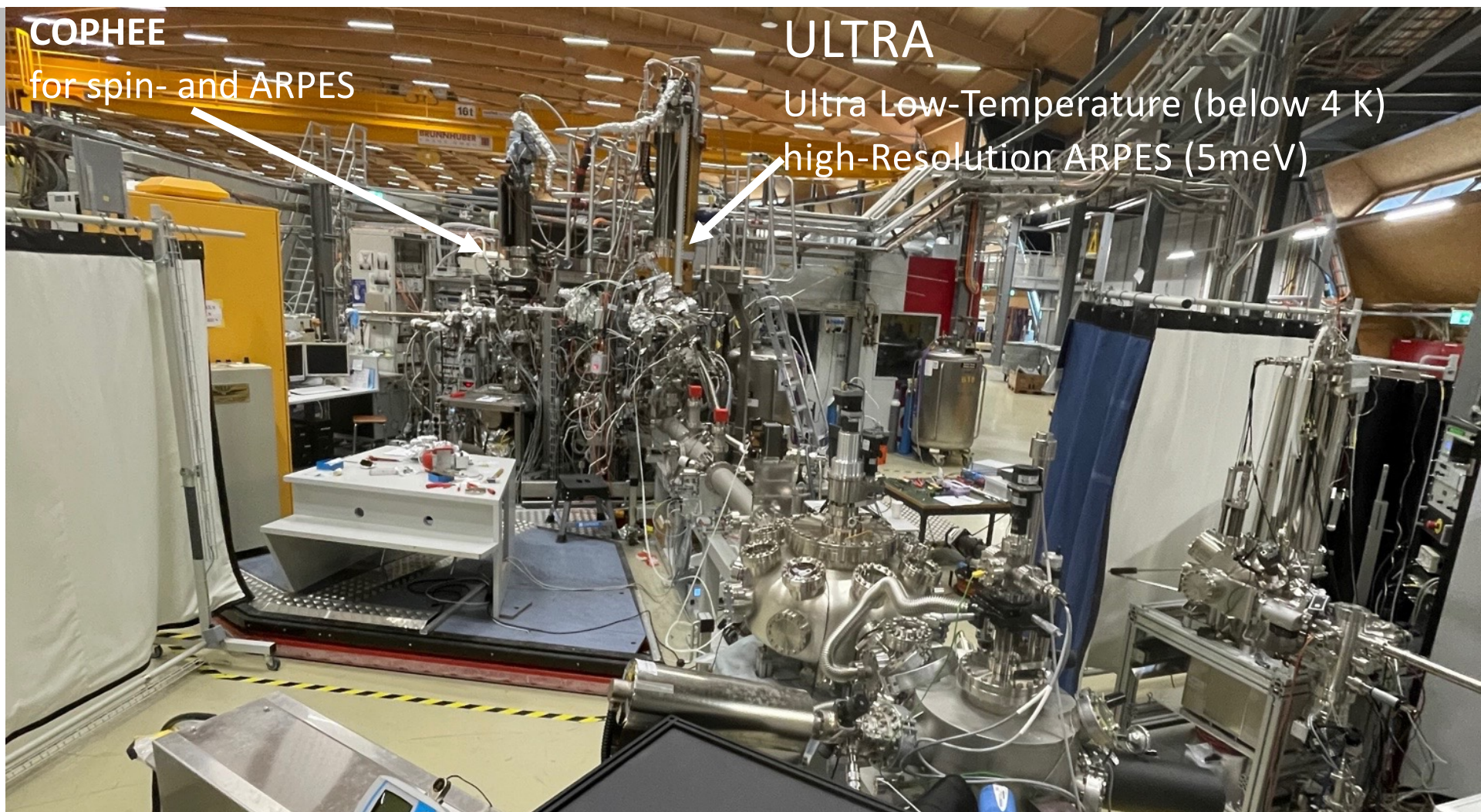
Designing novel functional materials with novel electrical, magnetic, thermal, chemical or electrochemical properties.



The modular system at the SIS beamline: ARPES+ PLD+STM+MBE



Two ARPES stations at SIS beamline:

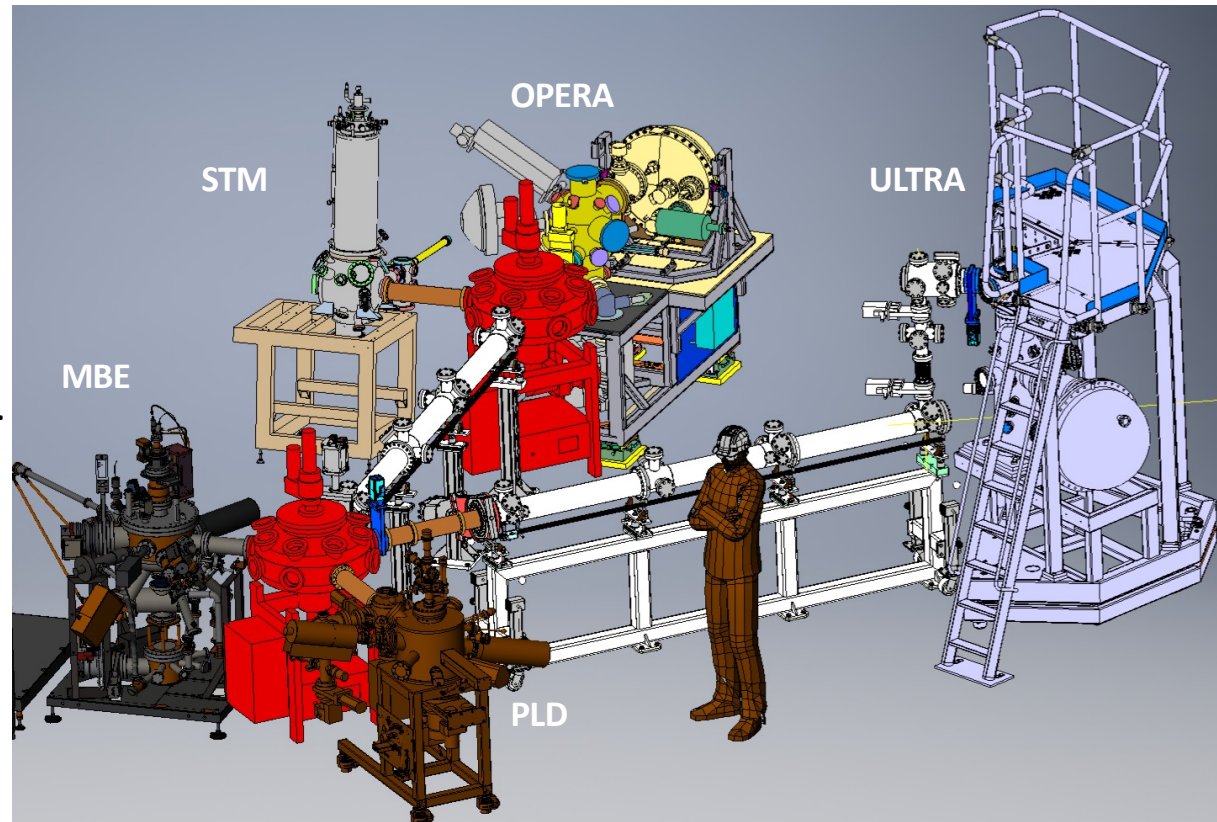


COPHEE
for spin- and ARPES

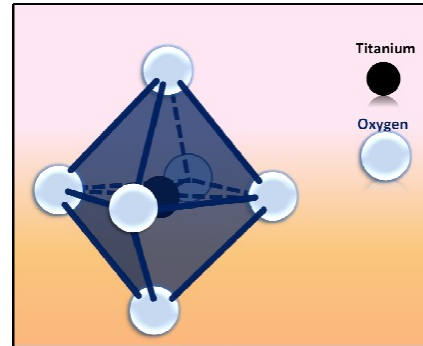
ULTRA
Ultra Low-Temperature (below 4 K)
high-Resolution ARPES (5meV)

QUEST (QUAntum matter Electron Spectroscopy Tool)

- 2 end stations – each utilizing **both** sources
 - **ULTRA** end station: low temperature, high resolution + spin detection.
 - **OPERA** end station: complex systems, operando, micro-focus.
- Advanced sample preparation.
methods: PLD, MBE.
- Complementary instrumentation
STS, STM, AFM.



Control of the electronic properties of ATiO_3



Distortion of the TiO_6 octahedron



Energy splitting and, the ordering of the d_{xy} & d_{xz} & d_{yz} bands (from one to multiple bands conductivity)

Doping (through A or O vacancies)



Filling of the bands (Carrier density, electron-phonon coupling..)

Octahedral rotations, Binding angles



Hopping probability (effective mass, band width, transport)

Collaborators

SIS beamline @ SLS:



Ming Shi



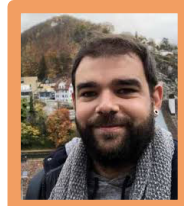
Nick Plumb



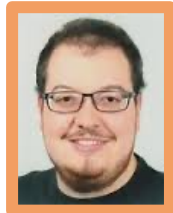
Hugo Dil (EPFL)



F. Baumberger
(PSI&U. Geneva)



E. Bonini Guedes



S. Muff



M. Naamneh



A. Chikina



Hang Li



J. Jandke



M. Caputo

ADRESS beamline @ SLS



Thorsten Schmitt



V. Strocov

PSI



M. Maedarde



Z. Salman



L. Patthey



J. Mesot

International

CNR-SPIN, Napoli, IT

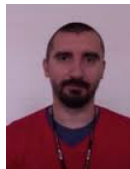


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CAS, China



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Jan Minar

DTU



Nini Pryds

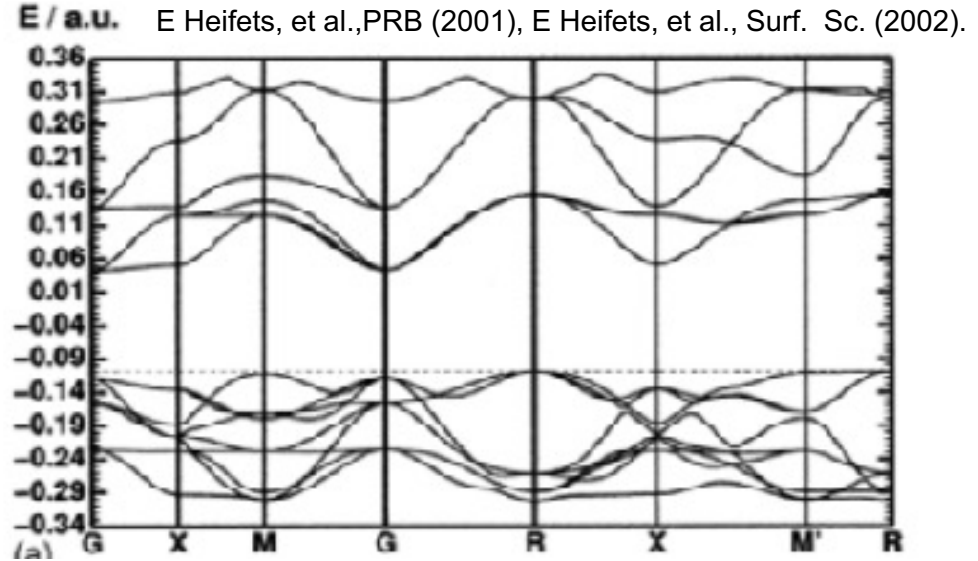
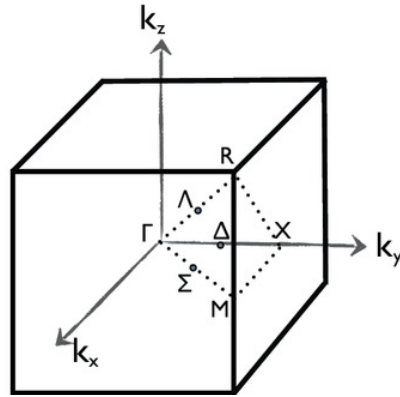
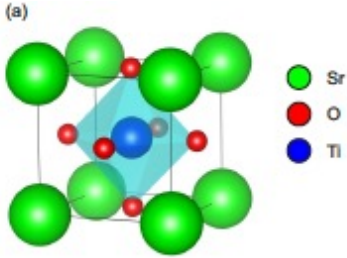
UFMG



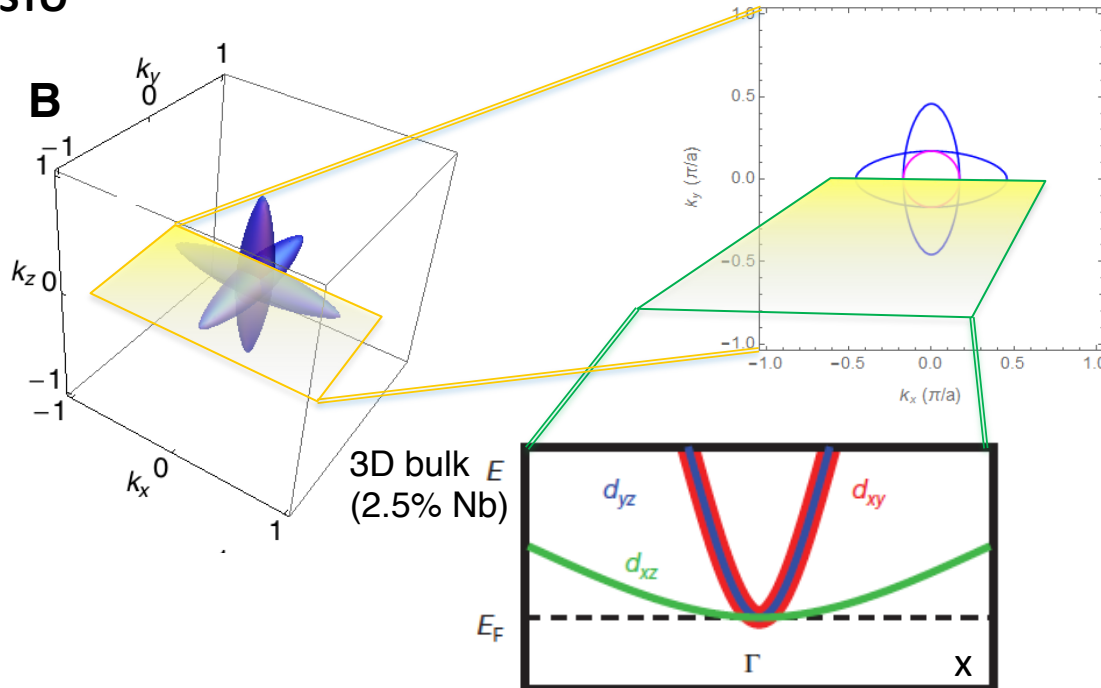
Walber Hugo de Brito

Band structure of SrTiO₃

- cubic above 105 K,
- slightly tetragonal below



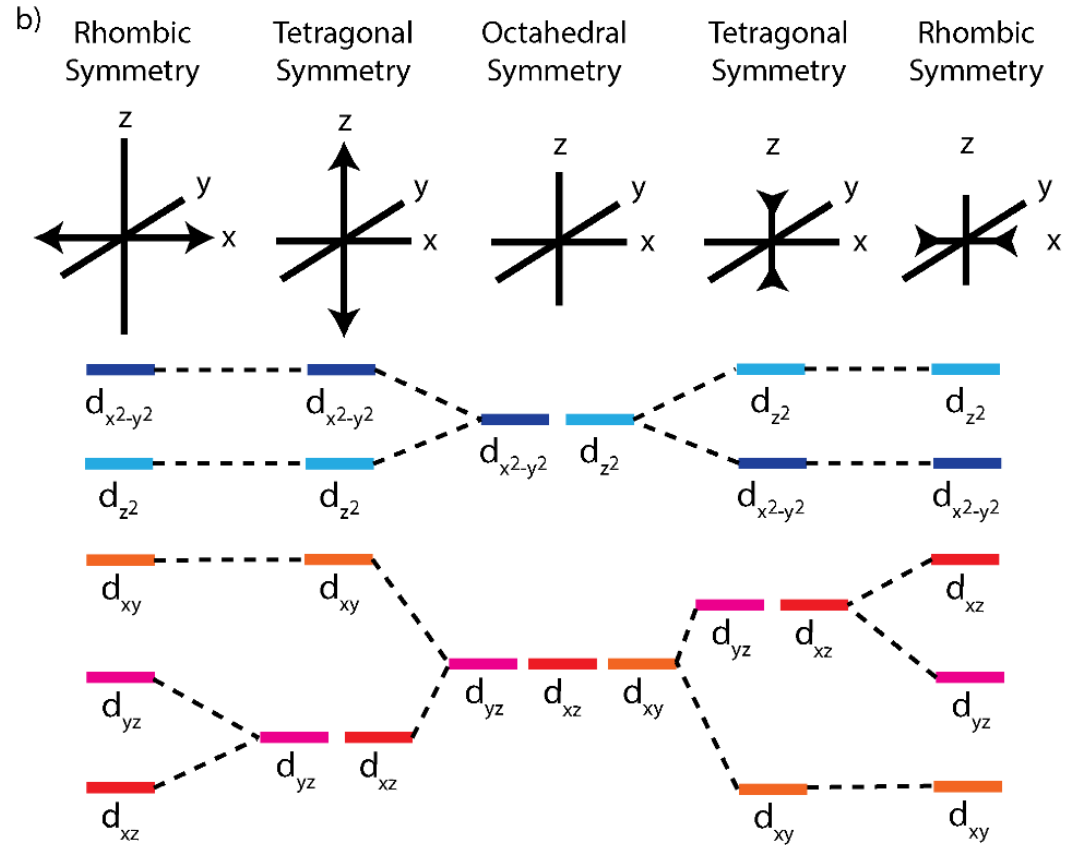
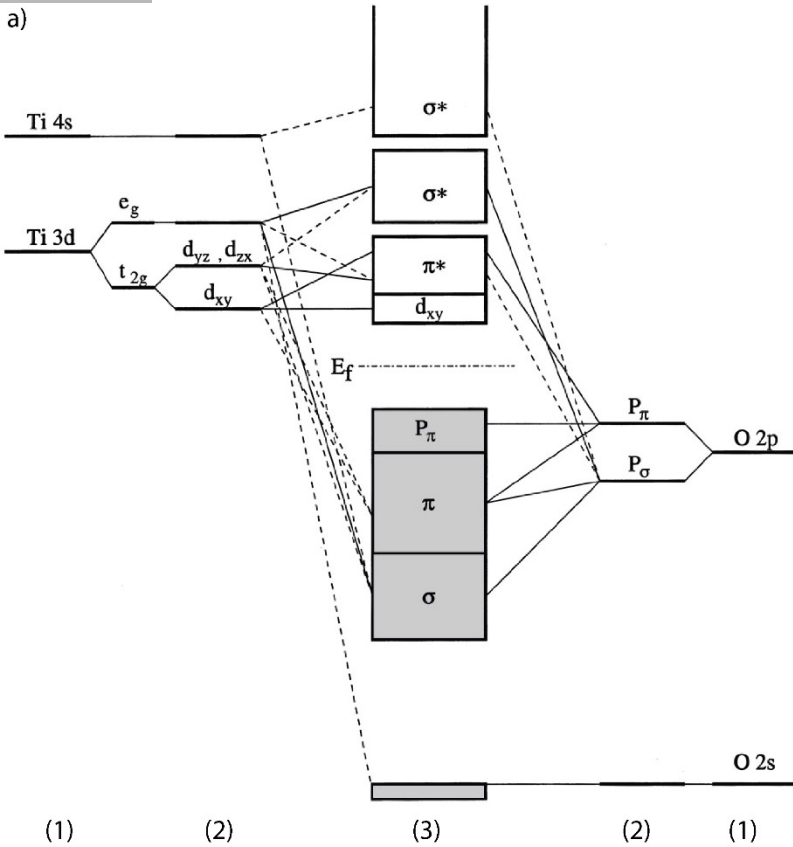
Fermi surface of doped STO



Orbital ordering of titanates

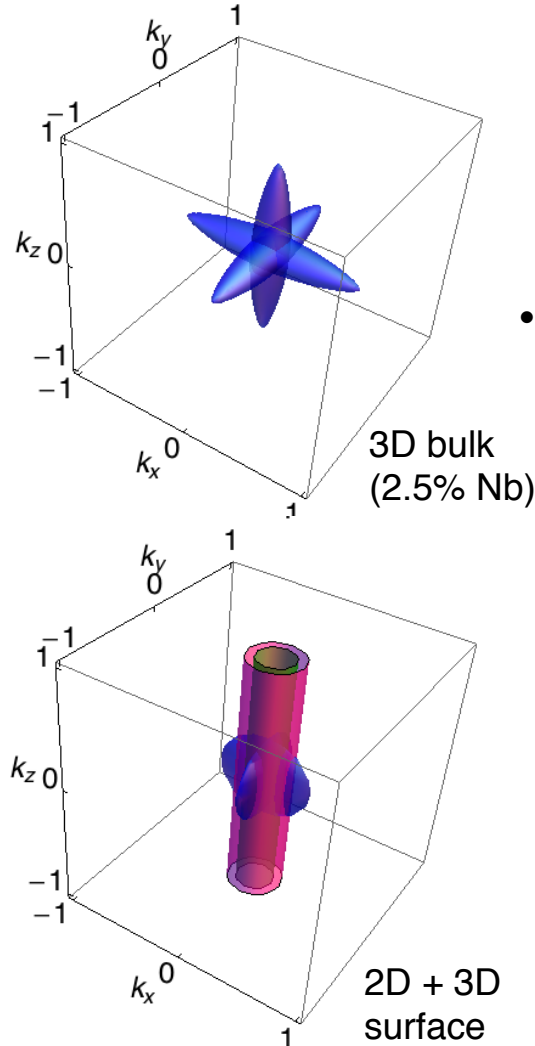
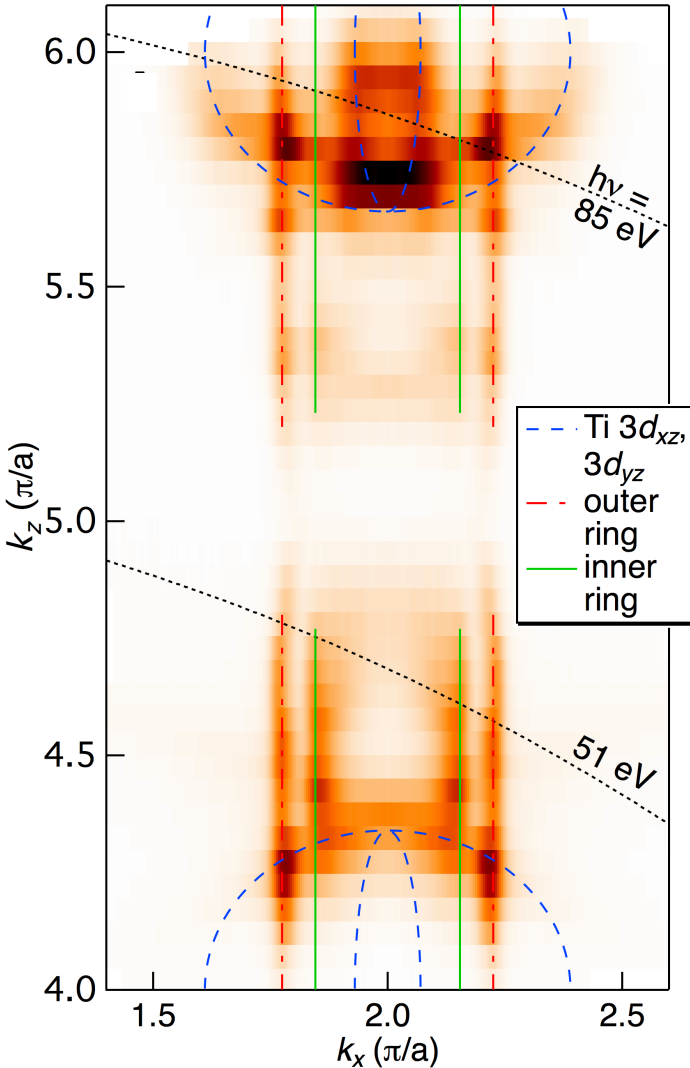
Tuning Orbital energies:

Splitting of the Ti 3d in t_{2g} and e_g determined by octahedra crystal field



Study of Metallic surface on STO: Surface-driven state with 3D dispersion

N.C. Plumb et al. PRL, 2014.

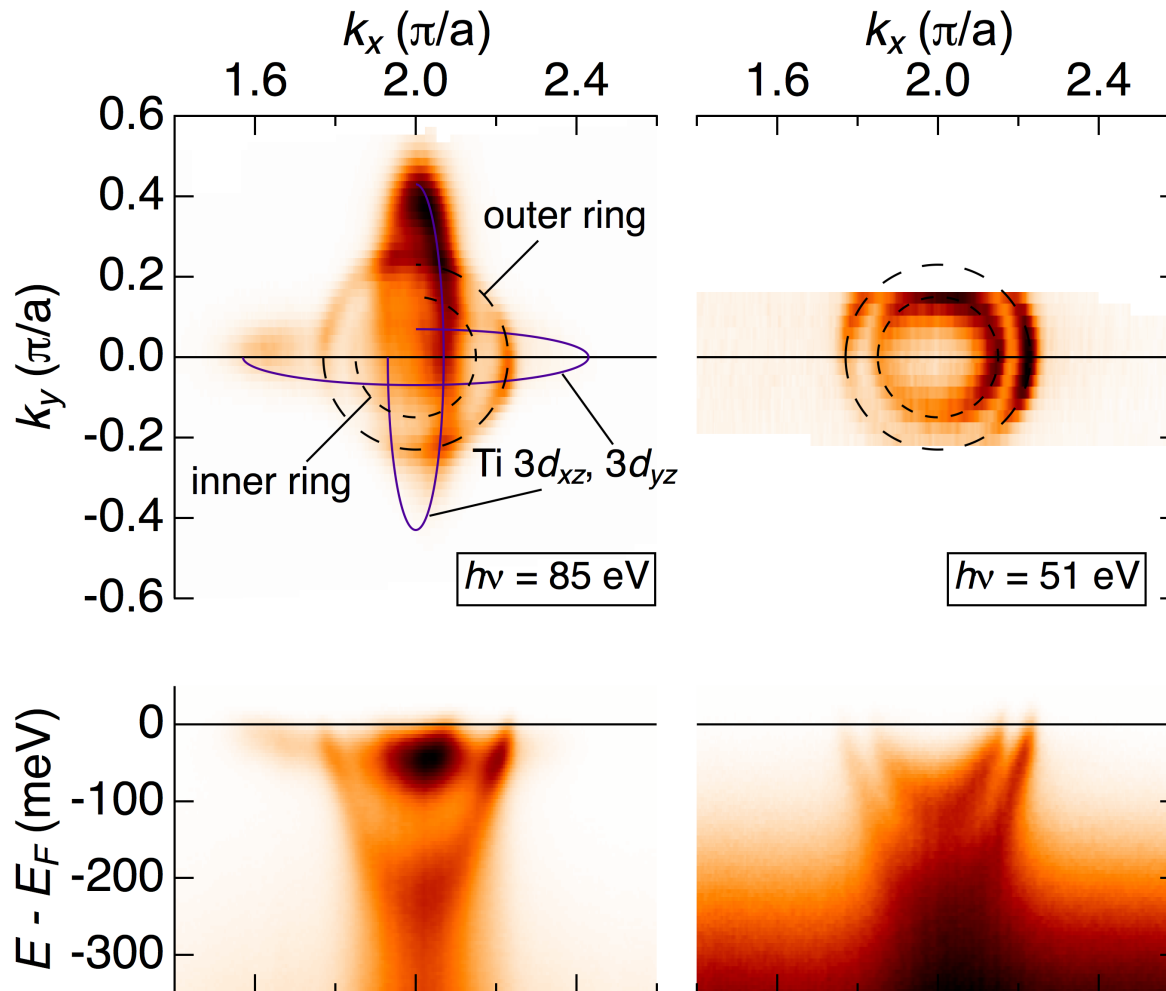


- All FS components differ drastically from bulk: From 3 prolate spheroids to:

- ❖ $d_{xy} \rightarrow$ 2D cylinder(s)
- ❖ $d_{xz}, d_{yz} \rightarrow$ 3D oblate ellipsoids

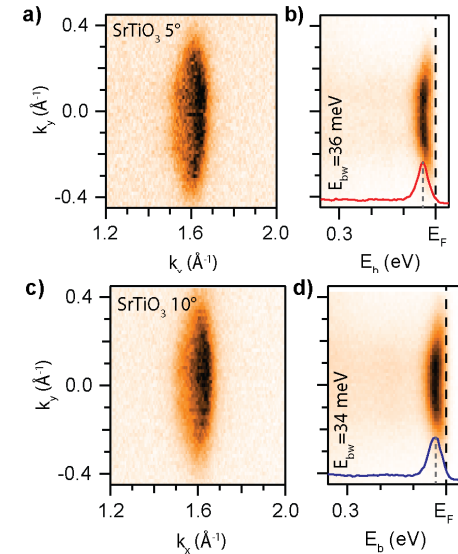
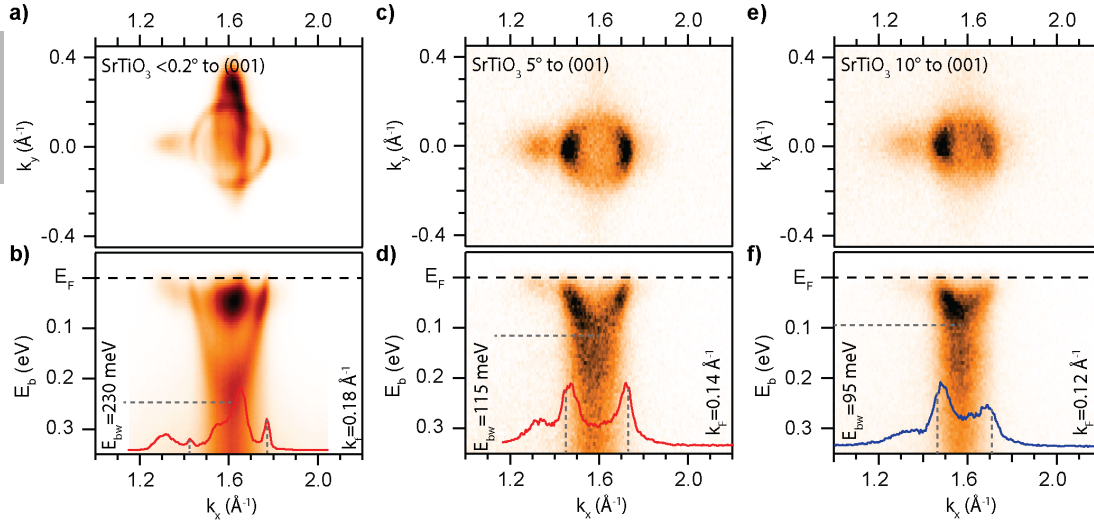
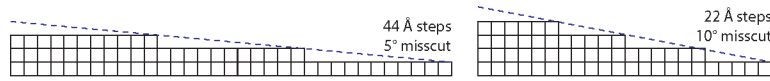
Study of Metallic surface on STO: Surface-driven state with 3D dispersion

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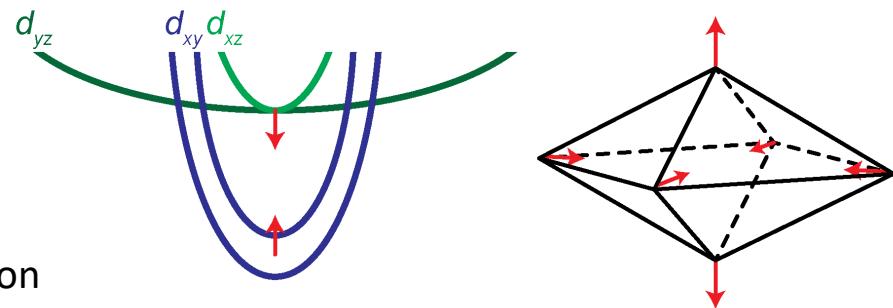
Terrace width controls the band splitting!

hν=85eV



Terrace size directly controls the two dimensional states:

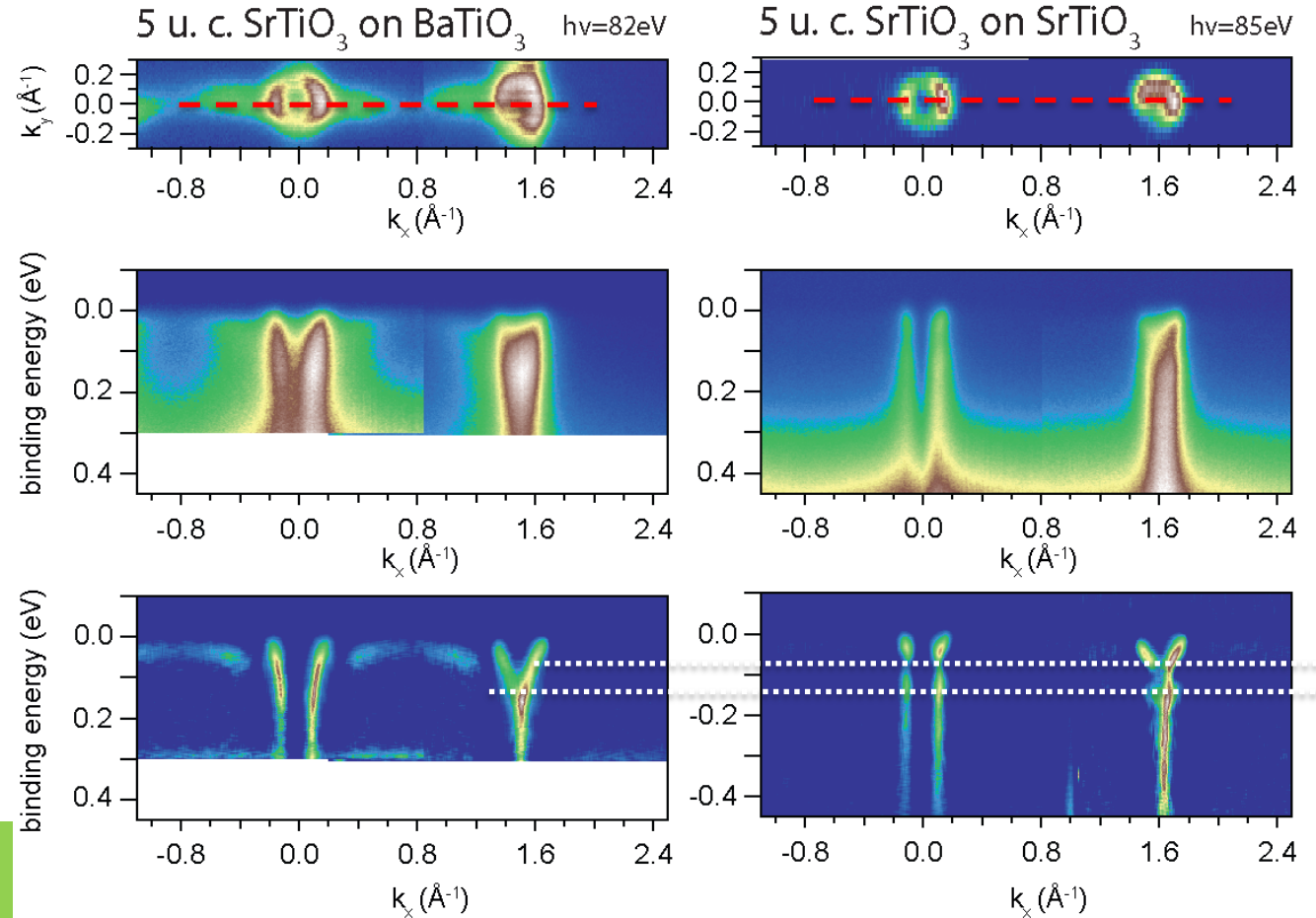
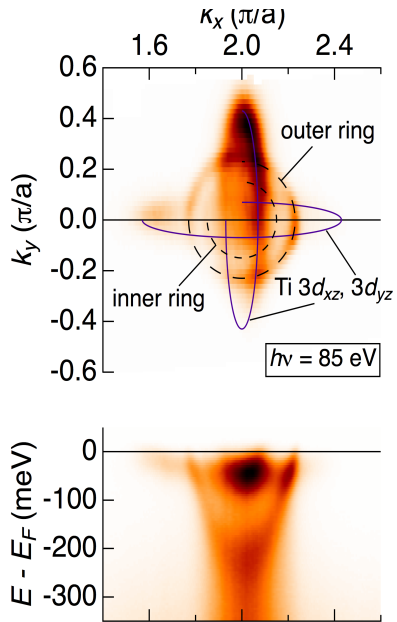
Flat SrTiO ₃	$m^* \approx 0.65 m_e$	$E_{\text{bandbot.}} \approx 235 \text{ meV}$
5° misscut	$m^* \approx 0.7 m_e$	$E_{\text{bandbot.}} \approx 110 \text{ meV}$
10° misscut	$m^* \approx 0.7 m_e$	$E_{\text{bandbot.}} \approx 90 \text{ meV}$



- altered octahedral distortion due to surface relaxation
- altered band filling due to changed electron affinity

An under layer controls the doping!

SrTiO₃ Wafer



5 u.c. SrTiO₃ on BaTiO₃

- 3d/heavy states formed
- $E_{\text{bandbottom}} \approx 150$ meV

5 u.c. SrTiO₃ on Nb:SrTiO₃

- No 3d/heavy bands, only d_{xy}
- $E_{\text{bandbottom}} \approx 60$ meV

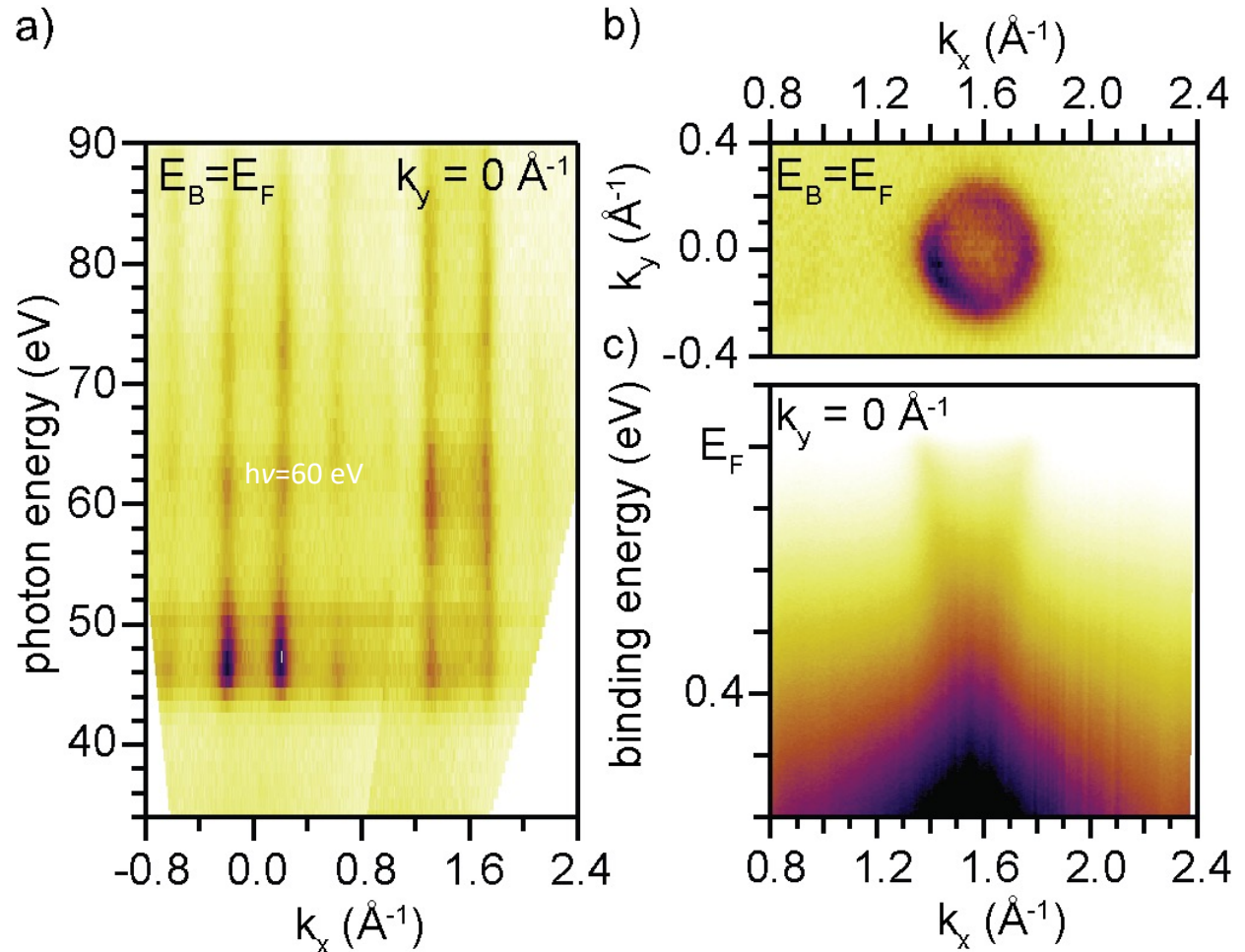
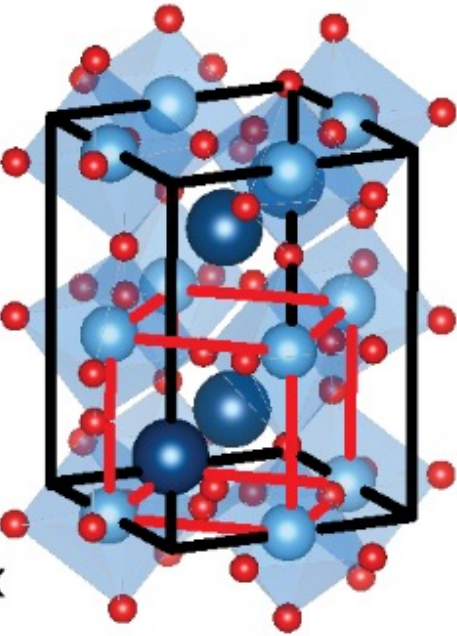
SrTiO₃

- d_{xy} and degenerated d_{xz}, d_{yz}
- $E_{\text{bandbottom}} \approx 240$ meV
- Band splitting ≈ 150 meV

N.C. Plumb et al. PRL, 2014.

Distortion and a rotation of the TiO₆ octahedron control the band splitting and filling

20 u.c. CaTiO₃ on Nb:SrTiO₃



Purely 2D electronic state

Only d_{xy} character (no d_{xz} , d_{yz})

$E_{bb} \approx 400 \text{ meV}$, $m^* = 0.4 m_e$

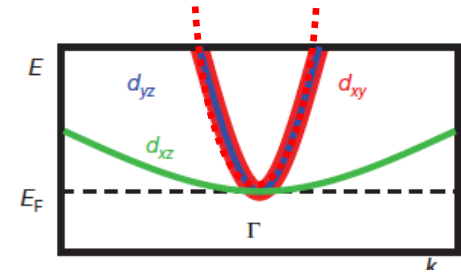
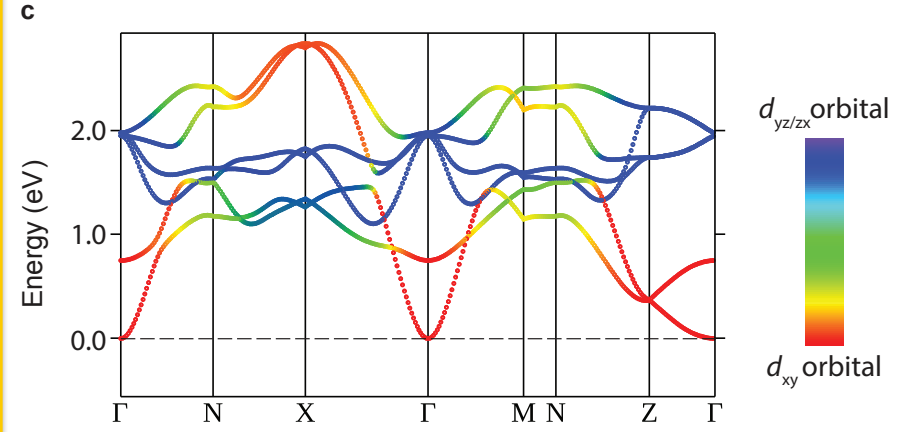
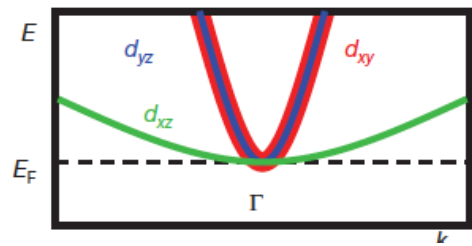
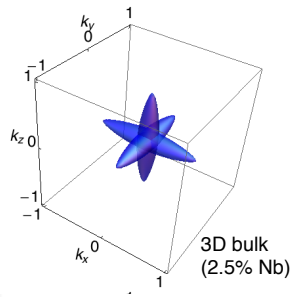
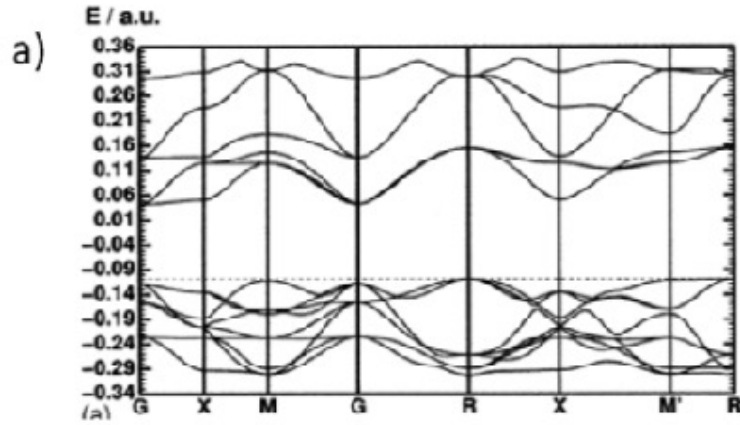
(SrTiO₃: $E_{bb} \approx 230 \text{ meV}$, $m^* = 0.7 m_e$)

From cubic to tetragonal TiO₂ octaedron...

Cubic (e.g. STO)



Tetragonal (e.g. anatase)



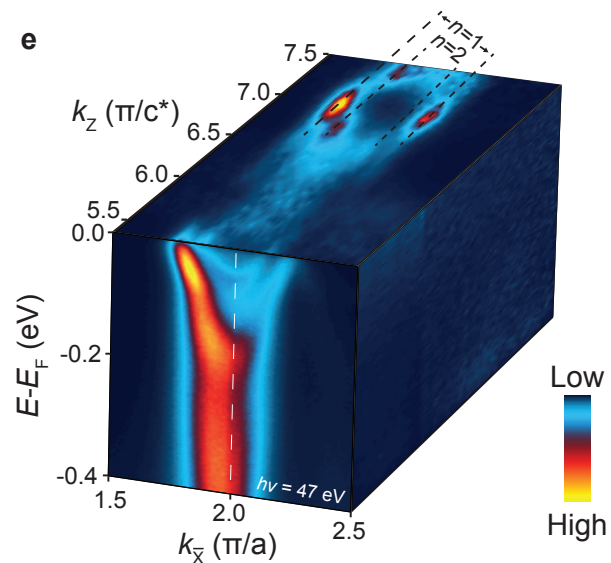
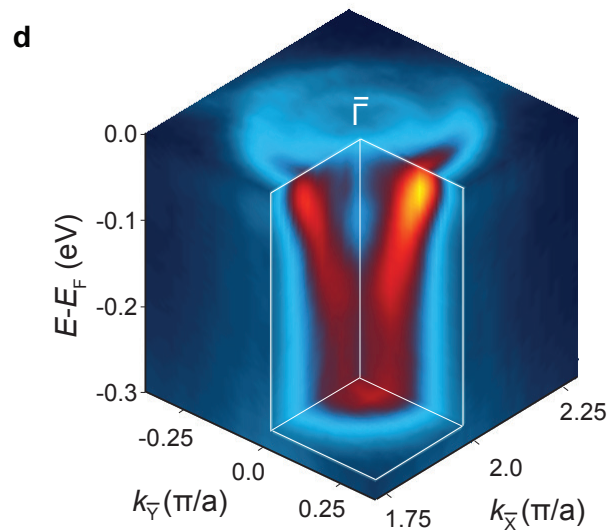
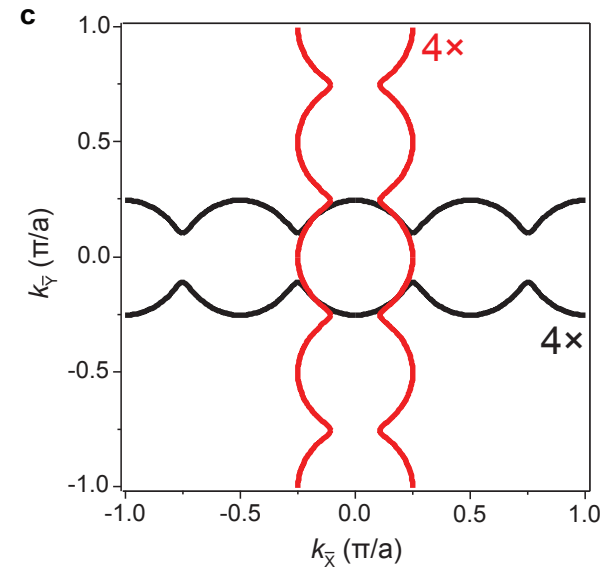
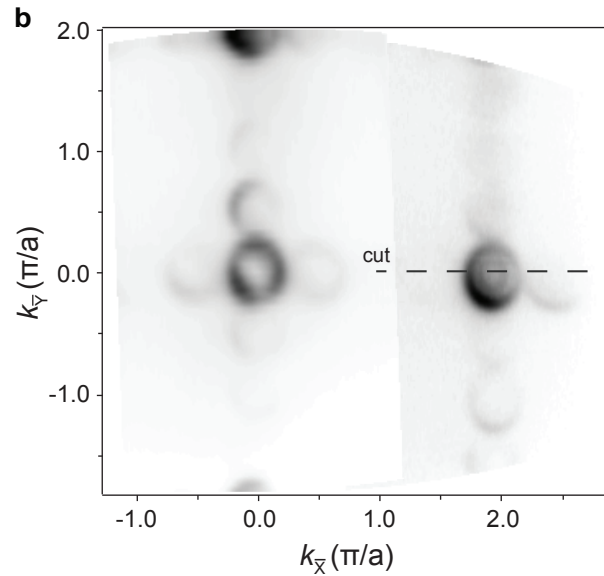
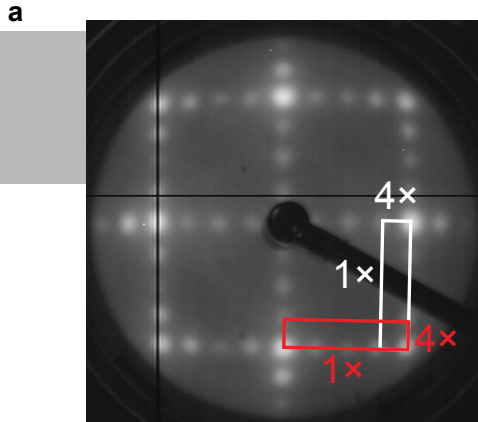
E Heifets, et al., PRB (2001),
E Heifets, et al. Surf. Sc. (2002).

Z. Wang, et al., Nano Letters, (2017).

ARPES on TiO₂ anatase film (20 u.c.)

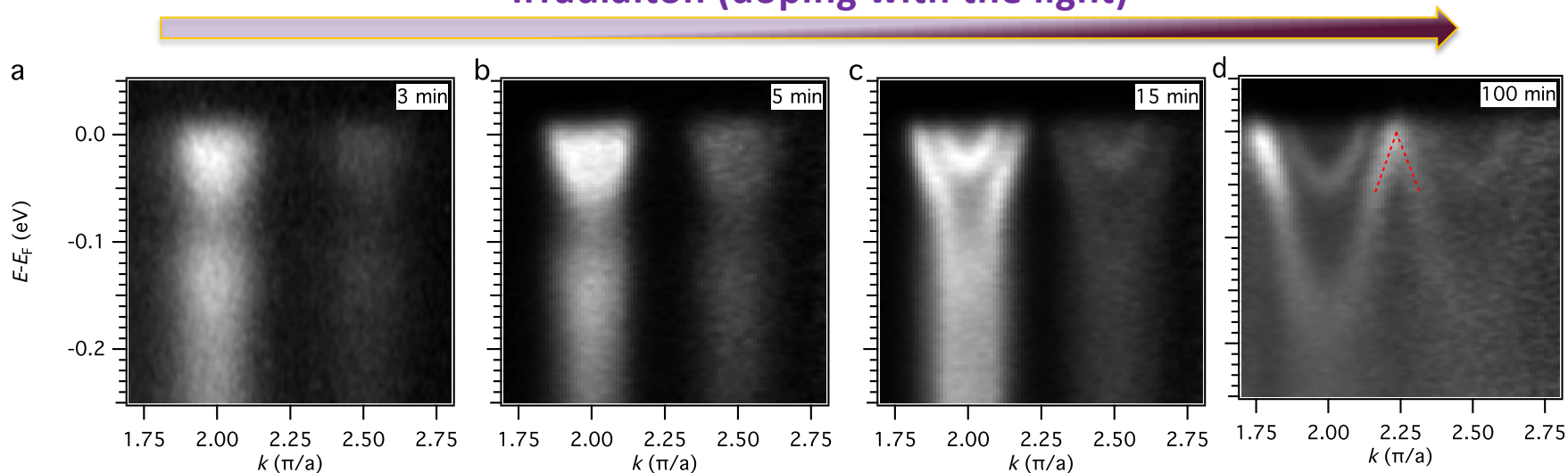
ARPES: 85 eV of the photon energy.

LEED

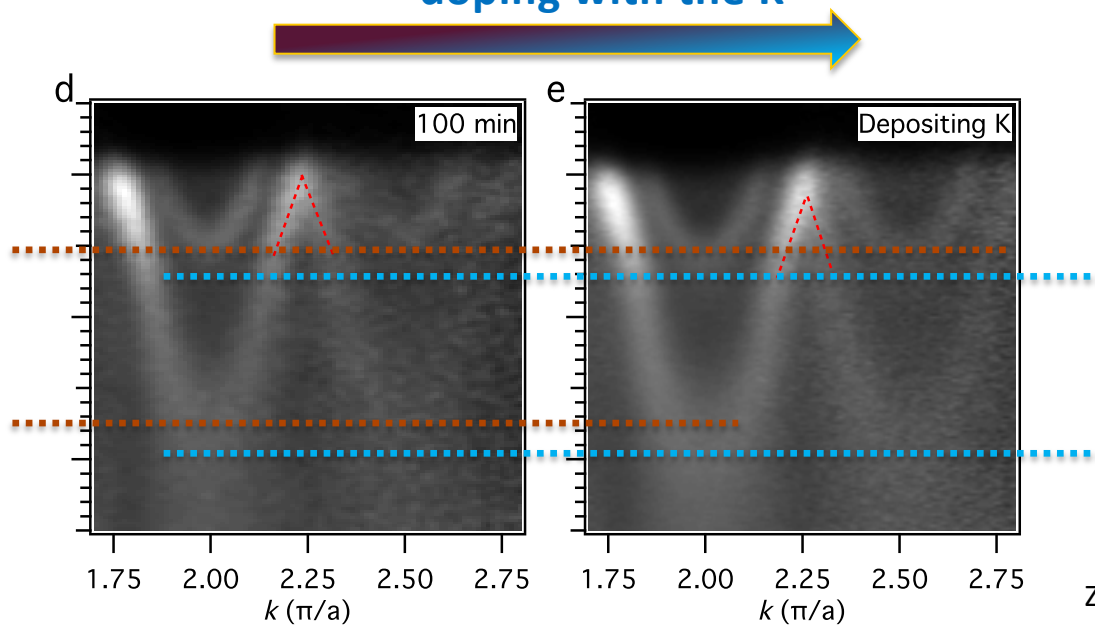


Control over the doping of anatase - TiO_2 film

Irradiation (doping with the light)



doping with the K



Take Home message :

Growth +ARPES is the powerful method for Engineering the electronic structure!

SrTiO₃ with different vicinality:

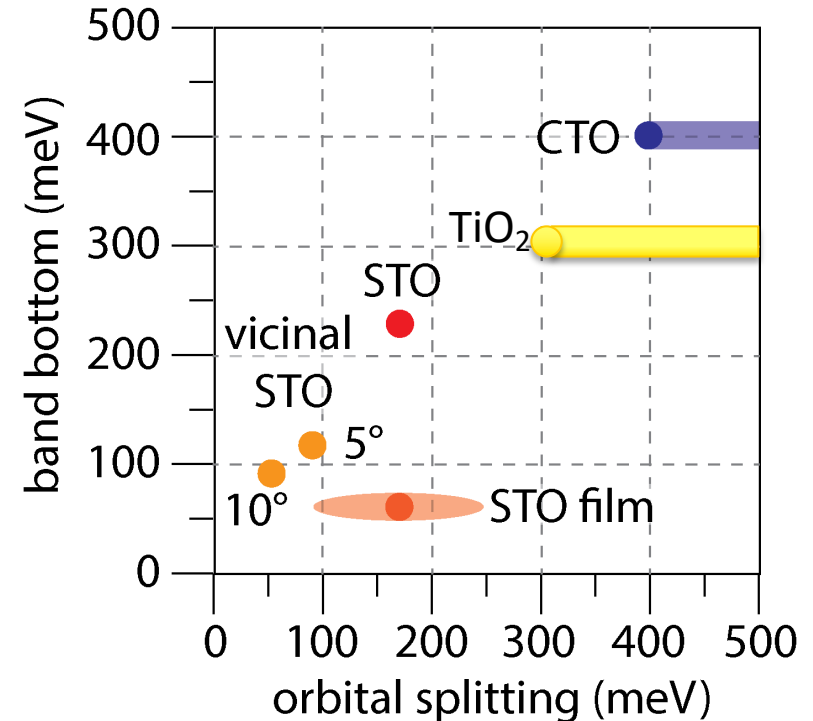
→ **Surface energy and distortion of octahedrons**

- ✓ reduced orbital splitting;
- ✓ reduced band filling;
- ✓ both also affected by temperature.

CaTiO₃ films:

→ Orthorhombic distortion of octahedrons

- ✓ increased orbital splitting
- ✓ increased band filling



SrTiO₃ films:

→ **defects**

- ✓ reduced band filling
 - ✓ altered band bending
- Single spin polarized band?

TiO₂ films:

tetragonal distortion of octahedrons

- ✓ large band filling
- altered band bending
- Spin polarized band?

- Phys. Rev. B. 86, 155425 (2012).
- Phys. Rev. Lett. 113, 086801 (2014).
- Nature Mat. 13, 1085–1090 (2014).
- Nature Mat. 15, 835–839 (2016).
- Nano Letters, 17 (4), pp 2561–2567 (2017).
- Applied Surface Science, 432A, (2018).
- Phys. Rev. B 98 (2018).
- Physical Review Research 2 (3) (2020).
- Advanced Science 8 (19), 2101516, (2021).